ASTR 1040 Recitation: White Dwarfs and Supernovae

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March 10 & 12, 2014

- Observing Session: Tonight Mar 10 (8:00 pm)
- MIDTERM: Thurs Mar 13 (regular class time, 9:30 am)

• Review Session: Wed Mar 12 (5:00 - 7:00 pm)

• Past / Current Homework Questions?

• White Dwarfs and Degeneracy Pressure

• Supernovae and Nuclear Reactions

"Basic" Quantum Mechanics

• Heisenberg Uncertainty Principle

Quantum

• Pauli Exclusion Principle

$$-\frac{\hbar}{2m}\nabla^{*}\Psi + V\Psi = i\hbar\frac{\partial\Psi}{\partial\epsilon}$$

• Planck's Constant: $h \approx 10^{-34}$ J s or $\hbar \equiv h/(2\pi)$ J s

"Basic" Quantum Mechanics

• Heisenberg Uncertainty Principle

- $\Delta x \Delta p \geq \hbar/2$
- $\Delta t \Delta E \geq \hbar/2$

• *p* is momentum and *E* is energy



Werner Heisenberg

"Basic" Quantum Mechanics

• Pauli Exclusion Principle

 No two fermions (protons, electrons, neutrons) can occupy the same quantum state

 Fermions have half-integer spin and Bosons (photons) have integer spin



Wolfgang Pauli

• White Dwarf: ~size of Earth, ~mass of Sun

• Supported by Electron Degeneracy Pressure

•
$$P_{\mathrm{NR}} = \frac{\hbar^2}{m_e} \left(\frac{Z}{A}\right)^{5/3} \left(\frac{\rho}{m_p}\right)^{5/3}$$



How did you get that result?



• Is there a relationship between Mass and Radius?

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• Set
$$P_{\rm HSE} = P_{\rm Deg} \quad \Rightarrow \quad R \propto M^{-1/3}$$

• We used P = nvp, what happens when $v \approx c$?

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• Simply replace $v \rightarrow c$

•
$$P_{\mathrm{R}} = \hbar c \left(\frac{Z}{A}\right)^{4/3} \left(\frac{\rho}{m_p}\right)^{4/3}$$

For
$$Z/A = 1$$
 and $\rho = 1$ g/cm⁻³

Non-Relativistic:

$$P_{
m NR}=9.9 imes10^{12}~
m dyn~cm^{-2}$$

Relativistic:

$$P_{\mathrm{R}} = 1.2 imes 10^{15} \mathrm{~dyn~cm^{-2}}$$

Unit conversions are good for the soul, so ...

Convert dyn cm⁻² (cgs) to SI/MKS unit of pressure: Pascal

Remember P = F/A and a dyn is cgs unit of force

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Convert dyn cm⁻² (cgs) to SI/MKS unit of pressure: Pascal Remember P = F/A and a dyn is cgs unit of force 1 dyn cm⁻¹ = 0.1 Pa Unit conversions are good for the soul, so ...

Convert dyn cm
$$^{-2}$$
 (cgs) to SI/MKS unit of pressure: Pascal

Remember P = F/A and a dyn is cgs unit of force

$$1 \text{ dyn cm}^{-1} = 0.1 \text{ Pa}$$

 $\begin{array}{l} 1 \text{ dyn } cm^{-2} = \frac{g \ cm}{s^2} \frac{1}{cm^2} \\ \\ \frac{g}{s^2 \ cm} = \frac{10^{-3} kg}{s^2 10^{-2} m} = 0.1 \ \frac{kg}{s^2 m} = 0.1 \ \frac{kg \ m}{s^2} \frac{1}{m^2} = 0.1 \ \text{Pa} \end{array}$

Classifying Supernovae – It's Complicated



R. Orvedahl (CU Boulder)

Supernova Onion Shell Burning



Why Stop at Iron (Z = 26)?



Naturally Occurring Elements with Z > 26 Exist!

- For high Z elements it is hard to get another charged particle close due to the high Coulomb potential barrier
- Not for neutrons: ${}^{A}_{Z}X + n \rightarrow {}^{A+1}_{Z}X + \gamma$

• Results in more massive nucleii that are stable or unstable against beta-decay: ${}^{A+1}_{Z}X \rightarrow {}^{A+1}_{Z+1}X + ?$

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•
$$_Z^{A+1}X \rightarrow _{Z+1}^{A+1}X + e^- + \bar{\nu}_e + \gamma$$

If beta-decay half-life is short compared to timescale for neutron capture

slow process or s-process reactions

tends to produce stable nucleii

If beta-decay half-life is long compared to timescale for neutron capture

rapid process or r-process reactions

tends to produce neutron rich nucleii

s-process tend to occur in normal phases of stellar evolution

r-process can occur during a supernova

Neither process plays a significant role in energy production

Accounts for abundances of nucleii with $A \gtrsim 60$, ($Z \gtrsim 26$)